

## NONISOTHERMAL EFFECTS DURING CO<sub>2</sub> LEAKAGE FROM A GEOLOGIC DISPOSAL RESERVOIR

Karsten Pruess

Contact: 510/486-6732, k\_pruess@lbl.gov

### RESEARCH OBJECTIVES

There is general consensus in the scientific community that geologic disposal of CO<sub>2</sub> into saline aquifers would be made at supercritical pressures,  $P > P_{crit} = 73.82$  bars. However, CO<sub>2</sub> escaping from a storage reservoir may migrate upwards towards regions with lower temperatures and pressures, where CO<sub>2</sub> would be in subcritical conditions. An assessment of the fate of leaking CO<sub>2</sub> requires the capability to model not only supercritical but also subcritical CO<sub>2</sub>, as well as phase changes between liquid and gaseous CO<sub>2</sub> in subcritical conditions.

### APPROACH

A new fluid property module was written for the general-purpose TOUGH2 simulator to represent mixtures of brine and CO<sub>2</sub> in all possible phase states, including gaseous, liquid, and supercritical CO<sub>2</sub>, as well as an aqueous phase and solid precipitate. This was applied to a study of leakage behavior in a simplified hypothetical model system.

### ACCOMPLISHMENTS

Starting from a typical geothermal gradient of 30°C/km in continental crust and hydrostatic pressures, the response to leaking CO<sub>2</sub> entering a vertical channel with elevated permeability at 1,000 m depth was simulated in 2-D cylindrical geometry. A first simulation case was run with an average land surface temperature of 5°C. In this case, the initial T,P-profile intersected the CO<sub>2</sub> saturation line, suggesting that liquid CO<sub>2</sub> will boil as it rises.

The system behavior observed in the simulation can be summarized as follows. Some of the CO<sub>2</sub> dissolved in water, but most of it formed a separate liquid phase. The liquid rose and started vaporizing at about a 630 m depth. Vaporization was partial and gave rise to evolution of a three-phase zone: aqueous—liquid CO<sub>2</sub>—gaseous CO<sub>2</sub>. Latent heat transfer during phase change caused considerable cooling of the rock, which allowed the liquid front to advance upward (see Figure 1). Over time, the

three-phase zone became several hundred meters thick. Upflow across the three-phase zone was impeded by interference between the phases. As a consequence, CO<sub>2</sub> discharge at the land surface was more dispersed than it otherwise would have been.

The simulation stopped after 391.2 years as freezing conditions were approached. This stoppage occurred because the fluid-property treatment adopted here has no provisions to deal with phase change from liquid water to ice or solid hydrate phases.

Another simulation was run for a larger land-surface temperature of 15°C, in which the initial T,P-profile did not intersect the CO<sub>2</sub> saturation line. Although the rising CO<sub>2</sub> was not subject

to phase change, there were nevertheless strong cooling effects because specific enthalpy of CO<sub>2</sub> increases upon depressurization. Over time, temperatures declined and a three-phase zone evolved, similar to what was observed for lower land-surface temperature.

### SIGNIFICANCE OF FINDINGS

Upward migration of CO<sub>2</sub> from a geologic disposal reservoir is accompanied by strong heat-transfer effects and gives rise to the evolution of a thick and broad three-phase zone.

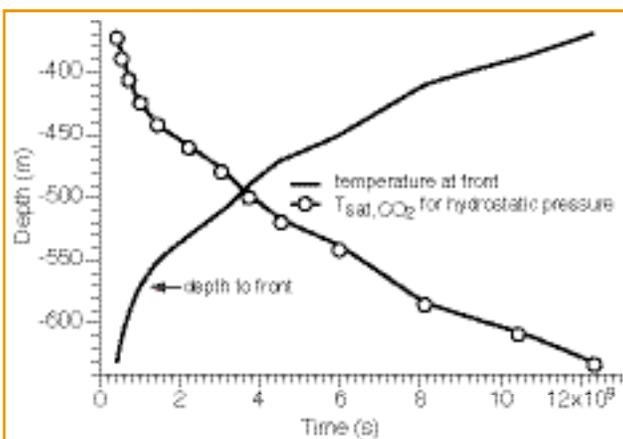


Figure 1. Advancement of liquid front, and frontal temperature, as a function of time

### RELATED PUBLICATION

Pruess, K., Numerical simulation of CO<sub>2</sub> leakage from a geologic disposal reservoir, including transitions from super- to subcritical conditions, and boiling of liquid CO<sub>2</sub>. SPE Journal, 2003 (submitted). Berkeley Lab Report LBNL-52423.

### ACKNOWLEDGMENTS

This work was supported by the Director, Office of Science, Office of Basic Energy Sciences, of the U.S. Department of Energy under Contract No. DE-AC03-76SF00098.